



(19) Europäisches Patentamt
European Patent Office
Office européen des brevets



(11) Publication number:

0 510 820 A2

(12)

EUROPEAN PATENT APPLICATION

(21) Application number: 92302746.0

(51) Int. Cl. 5: H01M 8/12, H01M 4/86

(22) Date of filing: 27.03.92

(30) Priority: 28.03.91 JP 87351/91

(71) Applicant: NGK INSULATORS, LTD.
2-56, Suda-cho, Mizuho-ku
Nagoya City Aichi Pref.(JP)

(43) Date of publication of application:
28.10.92 Bulletin 92/44

(72) Inventor: Seike, Shoji
167-1, Taishi 2-chome, Midori-Ku
Nagoya City, Aichi Pref.(JP)
Inventor: Yamada, Hirotake
66, Kawanayama-Cho 1-Chome, Showa-Ku
Nagoya City, Aichi Pref.(JP)

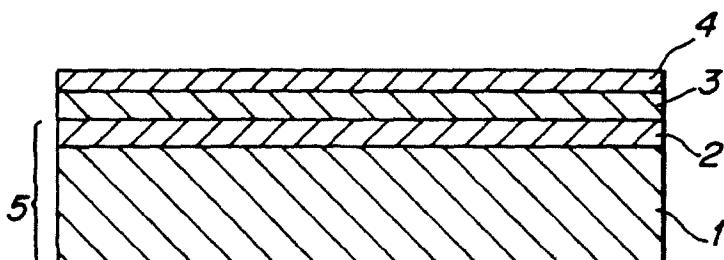
(84) Designated Contracting States:
BE DE FR GB

(74) Representative: Paget, Hugh Charles Edward
et al
MEWBURN ELLIS 2 Cursitor Street
London EC4A 1BQ(GB)

(54) Air electrode and solid electrolyte fuel cell having the same.

(57) An air electrode as a component of a solid electrolyte fuel cell. The air electrode is composed of a first layer (1) and a second layer (2). The first layer has an open porosity of 25% to 57%, pore diameters of 2.5 μ m to 12 μ m and a resistivity of less than 0.22 Ω cm. The second layer has an open porosity of 8% to 24%, pore diameters of 0.2 μ m to 3 μ m, and a ratio of the thickness of the second layer to the thickness of the air electrode is 2% to 28%. The sum of the thicknesses of the first and second layers is 0.7 mm to 3.0 mm. The materials of the first and second layers have perovskite structures selected from the group consisting of lanthanum manganate, calcium manganate, lanthanum nickelate, lanthanum cobalate and lanthanum chromate. A solid electrolyte fuel cell includes the air electrode as described above, a solid electrolyte film (3) formed on the surface of the second layer of the air electrode and a fuel electrode film (4) formed on the surface of the solid electrolyte film.

FIG. 1



This invention relates to an air electrode for a solid electrolyte fuel cell and more particularly to a solid electrolyte fuel cell having the air electrode.

Recently, fuel cells have been noted as power generating implements. The fuel cell is an implement capable of directly converting chemical energy included in a fuel to electric energy. As the fuel cell is free from limitation of Carnot's cycle, the cell is a very promising technique owing to its inherently high energy conversion efficiency, wide latitude of fuels to be used (naphtha, natural gas, methanol, coal reformed gas, heavy oil and the like), less public nuisance, and high electric power generating efficiency without being affected by scales of installations.

Particularly, as the solid electrolyte fuel cell (referred to as "SOFC" hereinafter) operates at high temperatures such as 1,000°C, activity of electrodes is very high. Moreover, the SOFC has low polarization and relatively high output voltage without requiring any catalyst of an expensive noble metal such as platinum so that energy conversion efficiency is much higher than those of the other fuel cells. Furthermore, the SOFC is stable and has long service life because all the constituent materials of the SOFC are solid.

In the SOFC, as the constituent components are solid, fuel cells of various structures have been proposed. Moreover, conductive perovskite type oxides of La(Sr)MnO₃ group, La(Sr)CoO₃ group and the like are expected as promising materials for the air electrode. Particularly, materials of La(Sr)MnO₃ group tend to match zirconia solid electrolyte films both in reactivity and thermal expansion coefficient.

However, the materials of the air electrodes described above are higher in electric resistance than materials of the fuel electrodes, which is the reason why output of the electric cells is comparatively low. Moreover, electric current flows along thin air electrode films or air electrode substrates and in parallel therewith, the electric resistance and voltage losses would become great as a whole.

In order to avoid this, it is conceivable to make the air electrode as thick as possible to reduce the electric resistance. With a thick air electrode, however, gas diffusion resistance in the air electrode becomes very high, with consequent considerable decrease of oxide agent supply amount to three-phase interfaces where the air electrode, the solid electrolyte film and the oxide agent contact one another. As a result, the output of the unit cell lowers contrary to the expectation.

In order to reduce the gas diffusion resistance in the air electrode to improve the permeability to the oxide agent, it is conceivable to make large the open porosity and pore diameters of the air electrode. However, such a countermeasure will reduce the number and areas of the three-phase interfaces where the air electrode, the solid electrolyte film and the oxide agent contact one another. Moreover, the larger the open porosity, the higher is the resistance of the electrode itself so that the output of the single cell lowers likewise.

It is an object of the invention to provide an air electrode and a solid electrolyte fuel cell having the air electrode, which are able to increase oxide agent supply amount to three-phase interfaces where the air electrode, the solid electrolyte film and the oxide agent contact one another and at the same time to increase the number and areas of the three-phase interfaces and lower the electric resistance in the air electrode, thereby increasing the output of the single cell.

In an air electrode as a component of a solid electrolyte fuel cell, according to the invention the air electrode comprises a first layer and a second layer, the first layer having an open porosity of 25% to 57%, pore diameters of 2.5 μm to 12 μm and a resistivity of less than 0.22 Ωcm, the second layer having an open porosity of 8% to 24%, pore diameters of 0.2 μm to 3 μm and a ratio of the thickness of the second layer to the thickness of the air electrode being 2% to 28%, and sum of the thicknesses of the first and second layers being 0.7 mm to 3.0 mm, and materials of the first and second layers have perovskite structures selected from a group consisting of lanthanum manganate, calcium manganate, lanthanum nickelate, lanthanum cobalate and lanthanum chromate.

Moreover, the solid electrolyte fuel cell according to the invention comprises the air electrode described above, a solid electrolyte film formed on the surface of the second layer of the air electrode and a fuel electrode film formed on the surface of the solid electrolyte film.

The invention resides in the discovery that the air electrode is formed by two layers taking partial charge of functions of the air electrode. In practice, one layer formed adjacent the solid electrolyte film is the second layer whose open porosity is 8% to 24% and pore diameters are 0.2 μm to 3 μm to increase the number and areas of the three-phase interfaces where the second layer, the solid electrolyte film and oxidizing agent contact one another.

Moreover, the ratio of the thickness of the second layer to the air electrode is less than 28% so that the relatively dense second layer restrains gas diffusion resistance from becoming excessively high. Further, as the ratio is more than 2%, it is possible to form sufficiently many three-phase interfaces.

Moreover, as the first layer has an open porosity more than 25% and pore diameters more than 2.5 μm, the gas diffusion resistance of the first layer is very low. Furthermore, as the first layer has an open

porosity of less than 57% and pore diameters less than 12 μm , the air electrode exhibits a high mechanical strength. Moreover, as the first layer has a resistivity of less than 0.22 Ωcm , the electric resistance of the whole air electrode becomes low to restrain electric voltage losses to an insignificant value.

With this arrangement, by combination of the respective components it is possible to elevate the output

5 of the fuel cell to the practically required output value which is more than 0.90 V. Moreover, in order to obtain the output value of more than 0.90 V, the sum of the thicknesses of the first and second layers should be within the range of 0.7 mm to 3.0 mm.

The invention will be more fully understood by referring to the following detailed specification and claims taken in connection with the appended drawings.

10 Fig. 1 is a sectional view conceptionally illustrating one example of the SOFC according to the invention; Fig. 2 is a conceptional view illustrating a preferred fine structure of the air electrode according to the invention; Fig. 3 is a graph illustrating the relation between the output voltage and the open porosity of the second layer in an experiment for determining the feature according to the invention; Fig. 4 is a graph illustrating the relation between the open porosity of the second layer and the distance from a plasma spray gun to the surface of the first layer in the experiment; Fig. 5 is a graph illustrating the relation between the output voltage and the ratio of the thickness of the second layer to the total thickness of the air electrode in an experiment for determining the feature according to the invention; Fig. 6 is a graph illustrating the relation between the thickness of the second layer and the number of plasma spray passes in the experiment; Fig. 7 is a graph illustrating the relation between the output voltage and pore diameters of the second layer in an experiment for determining the feature according to the invention; Fig. 8 is a graph illustrating the relation between the pore diameters of the second layer and granulated particle diameters of raw material for plasma spray in the experiment; Fig. 9 is a graph illustrating the relation between the output voltage and pore diameters of the first layer in an experiment for determining the feature according to the invention; Fig. 10 is a graph illustrating the relation between the pore diameters of the first layer and particle size of raw material for the first layer in the experiment; Fig. 11 is a graph illustrating the relation between the output voltage and pore diameters of the first layer in an experiment for determining the feature according to the invention; Fig. 12 is a graph illustrating the relation between the open porosity of the first layer and firing temperature of the first layer in the experiment; Fig. 13 is a graph illustrating the relation between the output voltage and the resistivity of the first layer in an experiment for determining the feature according to the invention; and Fig. 14 is a graph illustrating the relation between the amount of the pore increasing agent and the resistivity of the first layer in the experiment.

30 1 ... first layer
2 ... second layer
30 3 ... solid electrolyte film
4 ... fuel electrode film
40 5 ... air electrode film

Fig. 1 illustrates in section one example of the SOFC (single cell) according to the invention. The air electrode 5 is of a two-layer structure consisting of a first layer 1 and a second layer 2. A solid electrolyte film 3 is formed on the surface of the second layer 2. A fuel electrode film 4 is further formed on the surface of the solid electrolyte film 3. The shape of the flat plate-shaped SOFC in a plane is, for example, square, rectangular, circular or the like. Moreover, the air electrode according to the invention is also applicable to a so-called tubular SOFC or monolithic SOFC.

Fig. 2 conceptionally illustrates the fine structure of the air electrode 5 shown in Fig. 1. Respective particles sintered together in the first layer 1 are comparatively large, while those in the second layer 2 are comparatively small. The respective particles in the first layer 1 contact one another with wide areas to form as a whole a continuous skeleton. With such a fine structure, the contact surface areas of the particles in the first layer 1 are so large that the resistivity in the first layer 1 is small.

The first and second layers of the air electrode according to the invention have the following parameters determined to exhibit the practically required output voltage in various experiments later described.

The first layer has an open porosity of 25% to 57%, pore diameters of 2.5 μm to 12 μm and a resistivity of less than 0.22 Ωcm . The second layer has an open porosity of 8% to 24% and pore diameters of 0.2 μm to 3 μm , and a ratio of the thickness of the second layer to the thickness of the air electrode is

2% to 28%. The sum of the thicknesses of the first and second layers is 0.7 mm to 3.0 mm. Materials of the first and second layers have perovskite structures selected from a group consisting of lanthanum manganate, calcium manganate, lanthanum nickelate, lanthanum cobalate and lanthanum chromate.

5 The perovskite structures are preferably those whose lanthanum atoms or calcium atoms are partially stoichiometrically deficient. Moreover, it is preferable that the metal element included in the perovskite structures is partially substituted by an element selected from the group consisting of strontium, calcium, yttrium, cerium, ytterbium, zinc and barium.

10 The solid electrolyte fuel cell according to the invention comprises the air electrode as described above, a solid electrolyte film formed on the surface of the second layer of the air electrode and a fuel electrode film formed on the surface of the solid electrolyte film.

The solid electrolyte film 3 is preferably made of zirconia, and more preferably made of zirconia completely or partially stabilized with yttria, calcia, magnesia or ceria. The fuel electrode film 4 is preferably made of nickel-zirconia cermet, cobalt-zirconia cermet, chromium-zirconia cermet or the like.

15 Example

Flat plate-shaped solid electrolyte fuel cells shown in Fig. 1 were produced which were square in plane and of 20 mm × 20 mm size. Air electrodes 5 were made of $\text{La}_{0.9}\text{Sr}_{0.1}\text{MnO}_3$, whose thickness was 2 mm.

20 A raw material of $\text{La}_{0.9}\text{Sr}_{0.1}\text{MnO}_3$ was pulverized and its particle sizes were adjusted. Thereafter, the material was added with a pore increasing agent and mixed, and then formed into desired shapes by press-forming and sintered to form first layers 1. On the other hand, after the raw material of $\text{La}_{0.9}\text{Sr}_{0.1}\text{MnO}_3$ was pulverized and its particle sizes were adjusted, the material was coated on each of the first layers 1 by plasma spray coating to form a second layer 2. A solid electrolyte film 3 made of 8 mol yttria-stabilized zirconia (8YSZ) was provided by plasma spray coating on each of the second layers 2. The thickness of the 25 solid electrolyte film 3 was 100 m μ . Platinum paste was coated on the solid electrolyte film 3 by a brush to form a fuel electrode film 4.

The open porosity, pore diameters and resistivity of the first layer and the open porosity, pore diameters and thickness of the first layer were adjusted in a manner as later described. In this example, the fuel gas was hydrogen and the oxidizing gas was the air, while operating temperature was 1,000 °C.

30

(Experiment 1)

35 The open porosity of the second layer was changed within a range of 5% to 40% to search the open porosity range of the second layer where the output voltage was more than 0.90 V as shown in Fig. 3. In this case, the first layer had an open porosity of 40% and pore diameters of 8 μm and the ratio of the thickness of the first layer to the total thickness of the air electrode was 95%, while the second layer had pore diameters of 1 μm and the ratio of the thickness of the second layer to the total thickness of the air electrode was 5%. It had been found that the output voltages were more than 0.90 V within the range of 8% to 24% of the open porosity of the second layer, and the maximum output voltage was 0.97 V at the 15% 40 open porosity of the second layer.

The output voltages of the single cell were measured at current density of 250 mA/cm². This measurement condition of the output voltage was used likewise in the Experiments 2 to 6 later described.

45 The open porosity of the second layer was changed in the following manner. The second layer was formed by plasma spray coating described above. By changing distances (cm) between the plasma spray gun and the surface of the first layer in the plasma spray coating, the open porosity of the second layer was changed. The open porosity was measured by image analysis with a scanning electron microscope (SEM). In plasma spray coating, the output of the plasma spray was 35 KW and flow rates of gases were 10 l/min for hydrogen and 35 l/min for argon. Granulated particles of an average diameter of 40 μm were used as particles for the plasma spray Coating.

50

(Experiment 2)

55 The ratio of the thickness of the second layer to the thickness of the air electrode was changed within a range of 0% to 50% to search the open porosity range of the second layer where the output voltage was more than 0.90 V. In this case, the first layer had an open porosity of 40% and pore diameters of 8 μm , while the second layer had an open porosity of 15% and pore diameters of 1 μm . It had been found that the output voltages were more than 0.90 V within 2% to 28% ratio of the thickness of the second layer to the thickness of the air electrode and the maximum output voltage was 0.97 V at the 5% ratio.

The ratio of the thickness of the second layer was changed by varying the number of plasma spray coating passes to change the thickness of the second layer as shown in Fig. 6. At the same time, the first layer was ground to change its thickness so as to adjust the total thickness of the air electrode to be 2 mm. In plasma spray coating, the output of the plasma spray was 35 KW and flow rates of gases were 10 l/min for hydrogen and 35 l/min for argon. Granulated particles for the plasma spray coating had an average diameter of 40 μm . The distance between the plasma spray gun and the surface of the first layer was 15 cm.

(Experiment 3)

The pore diameters of the second layer were changed within a range of 0 μm to 8 μm to search the pore diameter range of the second layer where the output voltage was more than 0.90 V as shown in Fig. 7. In this case, the first layer had an open porosity of 40% and pore diameters of 8 μm , and the ratio of the thickness of the first layer to the total thickness of the air electrode was 95%, while the second layer had an open porosity of 15%, and the ratio of the thickness of the second layer to the total thickness of the air electrode was 5%. It had been found that the output voltages were more than 0.90 V within 0.2 μm to 3 μm of the pore diameters of the second layer, and the maximum output voltage was 0.97 V at 1 μm pore diameters of the second layer.

The pore diameter of the second layer was changed by varying granulated particle diameters of raw material for plasma spray coating as shown in Fig. 8. In plasma spray coating, the output of the plasma spray was 35 KW and flow rates of gases were 10 l/min for hydrogen and 35 l/min for argon. The distance between the plasma spray gun and the surface of the first layer was 15 cm.

(Experiment 4)

The pore diameters of the first layer were changed within a range of 1 μm to 14 μm to search the pore diameter range of the first layer where the output voltage was more than 0.90 V as shown in Fig. 9. In this case, the first layer had an open porosity of 40%, and the ratio of the thickness of the first layer to the total thickness of the air electrode was 95%, while the second layer had an open porosity of 15% and pore diameters of 1 μm , and the ratio of the thickness of the second layer to the total thickness of the air electrode was 5%. It had been found that the output voltages were more than 0.90 V within the range of the pore diameters of the first layer more than 2.5 μm , and the maximum output voltage was 0.97 V when the pore diameters of the first layer are more than 8 μm . Within the range of the pore diameters of the first layer more than 12 μm , the mechanical strength of the first layer was rapidly lowered.

The pore diameters of the first layer were adjusted by changing the particle size of the raw material of the first layer. In more detail, the raw material was pulverized to adjust the grain sizes and added with a pore increasing agent (acrylic powder). The material added with the pore increasing agent was formed into a desired shape by press-forming, which was then fired. As shown in Fig. 10, pore diameters of the first layers were changed by the particle sizes of the pulverized raw material. In this case, the firing temperature was changed within the range later described in order to maintain the open porosity of the first layer 40%. The pore increasing agent was 4 weight% and the firing temperatures were 1,250 °C to 1,680 °C.

(Experiment 5)

The open porosity of the first layer was changed within a range of 20 to 60% to search the open porosity range of the first layer where the output voltage was more than 0.90 V as shown in Fig. 11. In this case, the first layer had pore diameters of 8 μm , and the ratio of the thickness of the first layer to the total thickness of the air electrode was 95%, while the second layer had an open porosity of 15% and pore diameters of 1 μm , and the ratio of the thickness of the second layer to the total thickness of the air electrode was 5%. It had been found that the output voltages were more than 0.90 V within the range of open porosity of 25% to 57%, and the maximum output voltage was 0.9 V at 40% open porosity of the first layer.

In order to change the open porosity of the first layer, the formed bodies were fired at various temperatures as shown in Fig. 12. The pore increasing agent was 4 weight% and granulated particle diameters before the press-forming were 13 μm .

(Experiment 6)

The resistivity of the first layer was changed within a range of $0.2 \Omega\text{cm}$ to $0.28 \Omega\text{cm}$ to search the resistivity of the first layer where the output voltage was more than 0.9 V as shown in Fig. 13. In this case, the first layer had an open porosity of 40% and pore diameters of $8 \mu\text{m}$, and the ratio of the thickness of the first layer to the total thickness of the air electrode was 95%, while the second layer had an open porosity of 15% and pore diameters of $1 \mu\text{m}$, and the ratio of the thickness of the second layer to the total thickness of the air electrode was 5%. It had been found that the output voltage was more than 0.90 V within the range of the resistivity of less than $0.22 \Omega\text{cm}$.

The resistivity of the first layer was varied by changing amounts of the pore increasing agent (acrylic powder) as shown in Fig. 14. By changing the amounts of the pore increasing agent, the bulk density of the first layer was changed and the firing temperature was changed within $1,200^\circ\text{C}$ to $1,680^\circ\text{C}$. However, the open porosity of the first layer was fixed at 40%. By changing the firing temperature, growing degrees of connections (necks) between adjacent particles were changed to vary contact areas between the adjacent particles, with the result that the resistivity of the first layer was changed correspondingly thereto.

In this Experiment, the pore increasing agent was added to the raw material of the first layer in order to relatively enlarge the open porosity and the pore diameters of the first layer. However, in order to adjust the open porosity and pore diameters of the fired first layer, a plate-shaped body formed by press-forming may be worked to form apertures before firing or after firing.

(Experiment 7)

The experiment was effected to search the relation between the sum of thicknesses of the first and second layers and output of the single cell. The first layer had an open porosity of 40% and pore diameters of $8 \mu\text{m}$, and the ratio of the thickness of the first layer to the total thickness of the air electrode was 95%, while the second layer had an open porosity of 15% and pore diameters of $1 \mu\text{m}$, and the ratio of the second layer to the total thickness of the air electrode was 5%. The thickness of the second layer was varied by changing the number of the plasma spray coating passes. At the same time, the thickness of the first layer was changed by grinding it to adjust the ratio between the thicknesses of the first and second layers to be constant.

The sum of the thicknesses of the first and second layers was changed as follows and measured the output voltages of the unit cell.

	Sum of thicknesses of first and second layers (mm)	Output voltage (V)
35	0.7	0.90
	1.0	0.92
	1.5	0.95
	2.0	0.97
	2.5	0.93
	3.0	0.90
45	0.5	0.87
	5.0	0.83

As can be seen from the results, the practically required output voltage of more than 0.90 V could be obtained within the range 0.7 mm to 3.0 mm of the sum of thicknesses of the first and second layers.

In the air electrode and the solid electrolyte fuel cell having the air electrode according to the invention, it is possible to increase oxide agent supply amount to three-phase interfaces where the air electrode, the solid electrolyte film and the oxide agent contact one another and at the same time to increase the number and areas of the three-phase interfaces and lower the electric resistance in the air electrode, thereby increasing the output of the fuel cell as described above.

While the invention has been particularly shown and described with reference to preferred embodiments thereof, it will be understood by those skilled in the art that the foregoing and other changes in form and details can be made therein without departing from the spirit and scope of the invention.

Claims

1. An air electrode as a component of a solid electrolyte fuel cell, wherein said air electrode comprises a first layer and a second layer, said first layer having an open porosity of 25% to 57%, pore diameters of 2.5 μm to 12 μm and a resistivity of less than 0.22 Ωcm , said second layer having an open porosity of 8% to 24%, pore diameters of 0.2 μm to 3 μm and a ratio of the thickness of the second layer to the thickness of the air electrode being 2% to 28%, and sum of the thicknesses of the first and second layers being 0.7 mm to 3.0 mm, and materials of the first and second layers have perovskite structures selected from a group consisting of lanthanum manganate, calcium manganate, lanthanum nickelate, lanthanum cobalate and lanthanum chromate.
2. The air electrode as set forth in claim 1, wherein said perovskite structures are those whose lanthanum atoms are partially stoichiometrically deficient.
- 15 3. The air electrode as set forth in claim 1, wherein said perovskite structures are those whose calcium atoms are partially stoichiometrically deficient.
- 20 4. The air electrode as set forth in claim 1, wherein the metal element included in the perovskite structures is partially substituted by an element selected from the group consisting of strontium, calcium, yttrium, cerium, ytterbium, zinc and barium.
- 25 5. A solid electrolyte fuel cell comprising the air electrode as set forth in claim 1, a solid electrolyte film formed on the surface of the second layer of the air electrode and a fuel electrode film formed on the surface of the solid electrolyte film.

25

30

35

40

45

50

55

FIG. 1

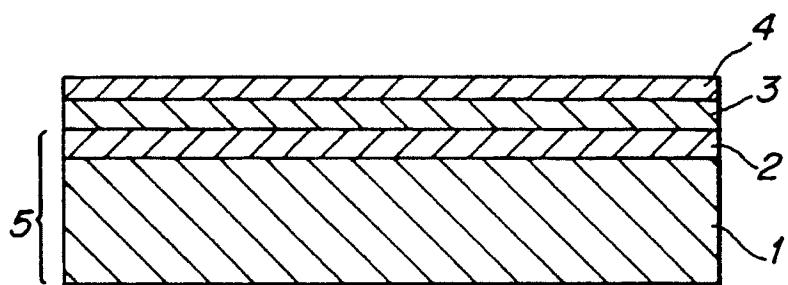


FIG. 2

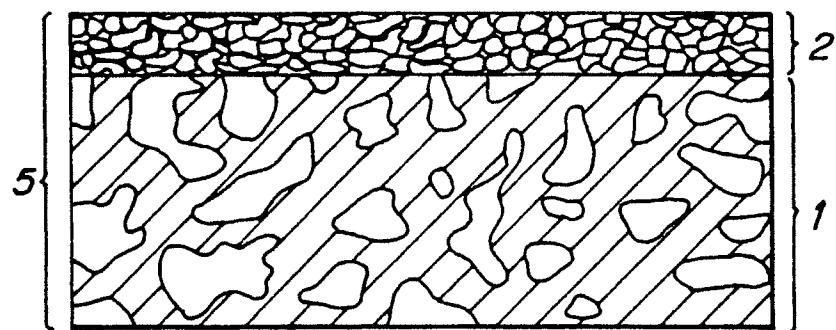


FIG.3

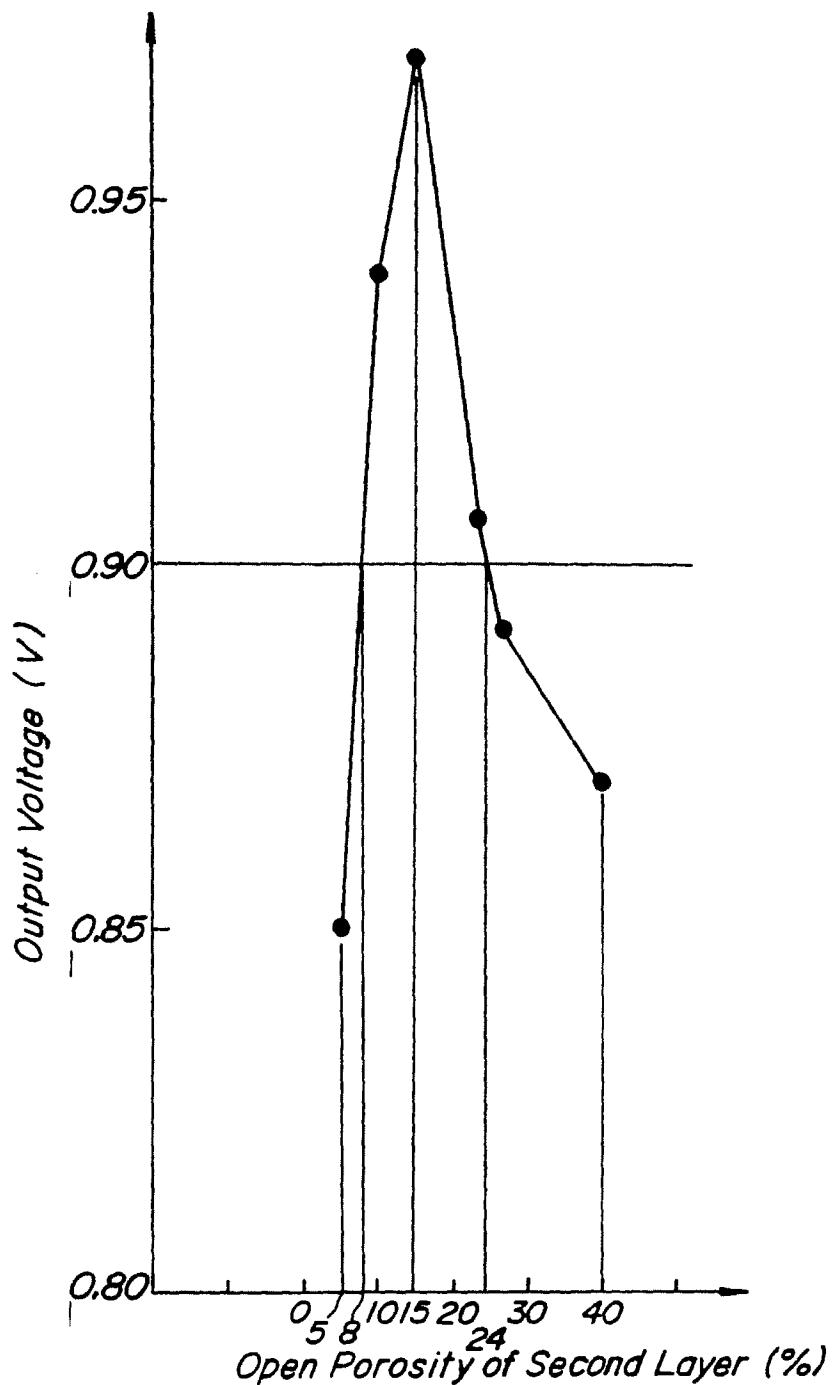


FIG. 4

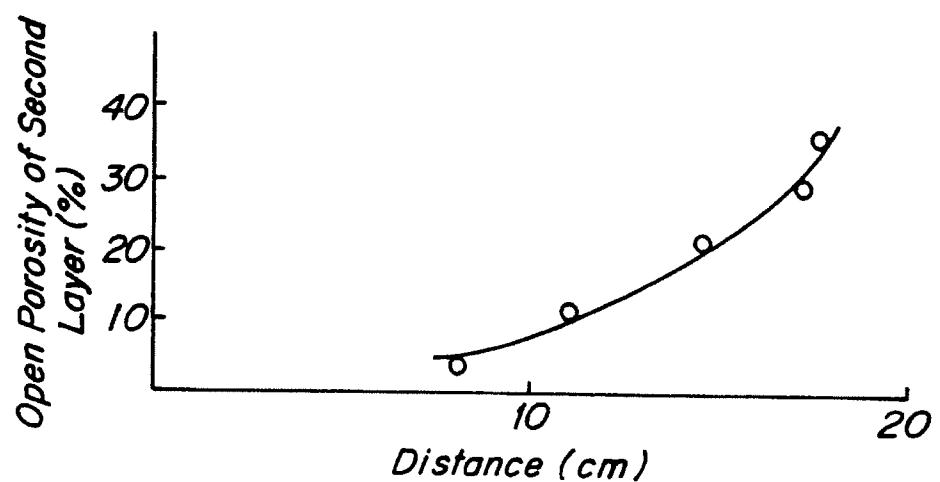


FIG. 5

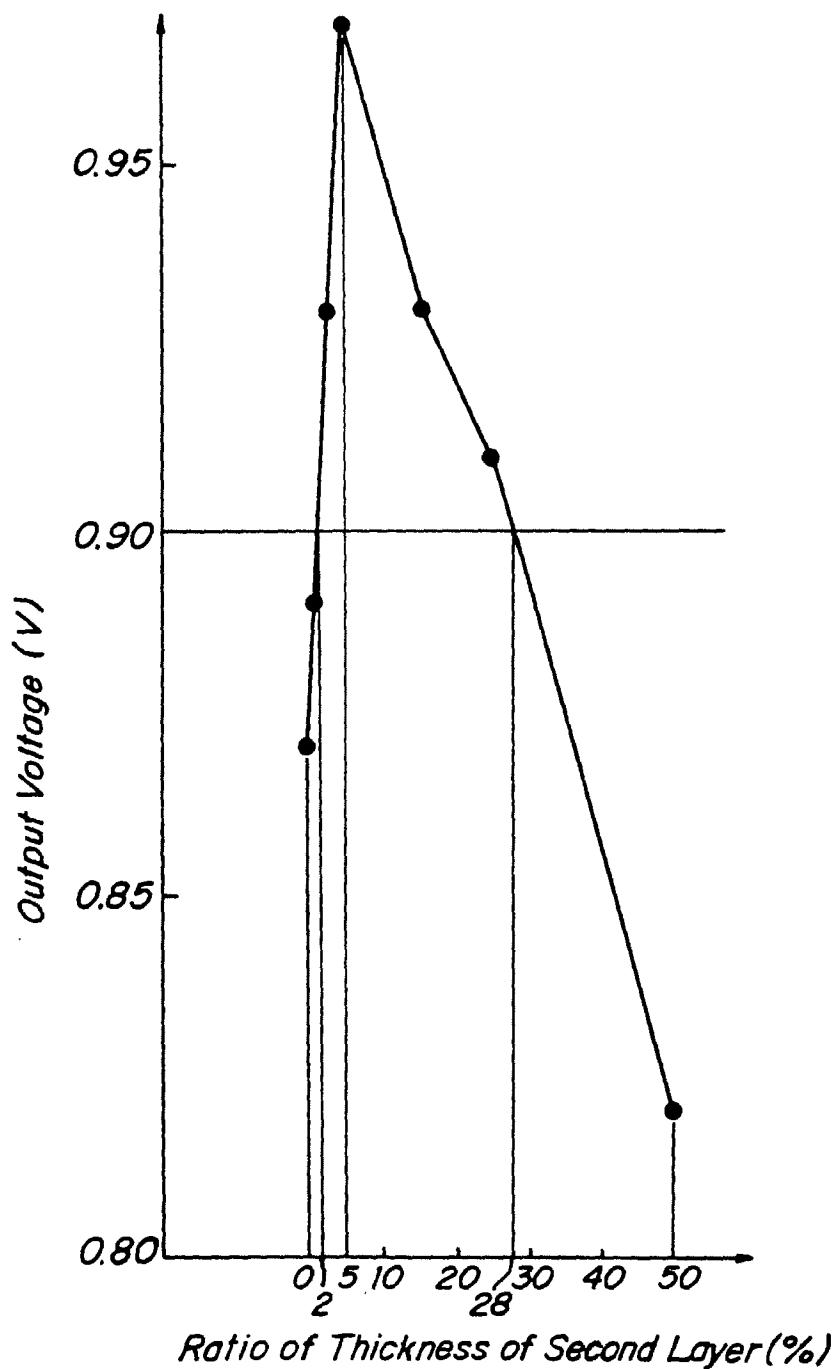


FIG.6

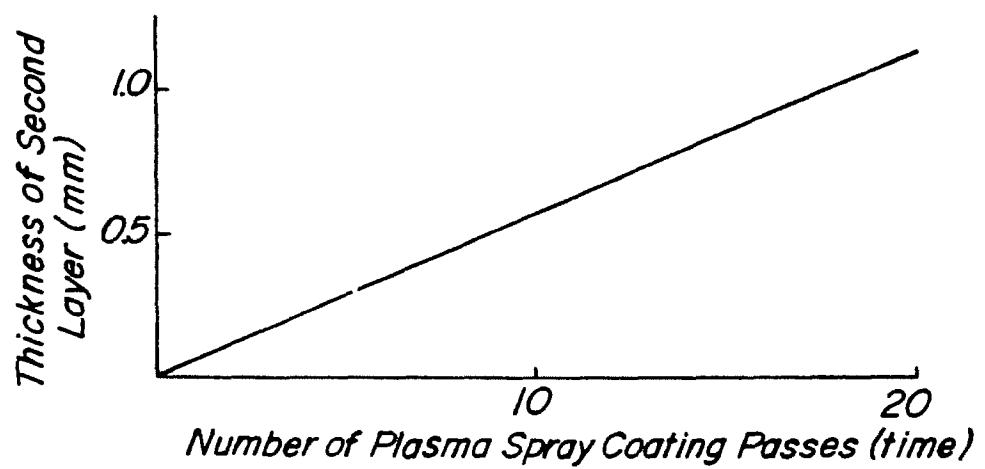


FIG. 7

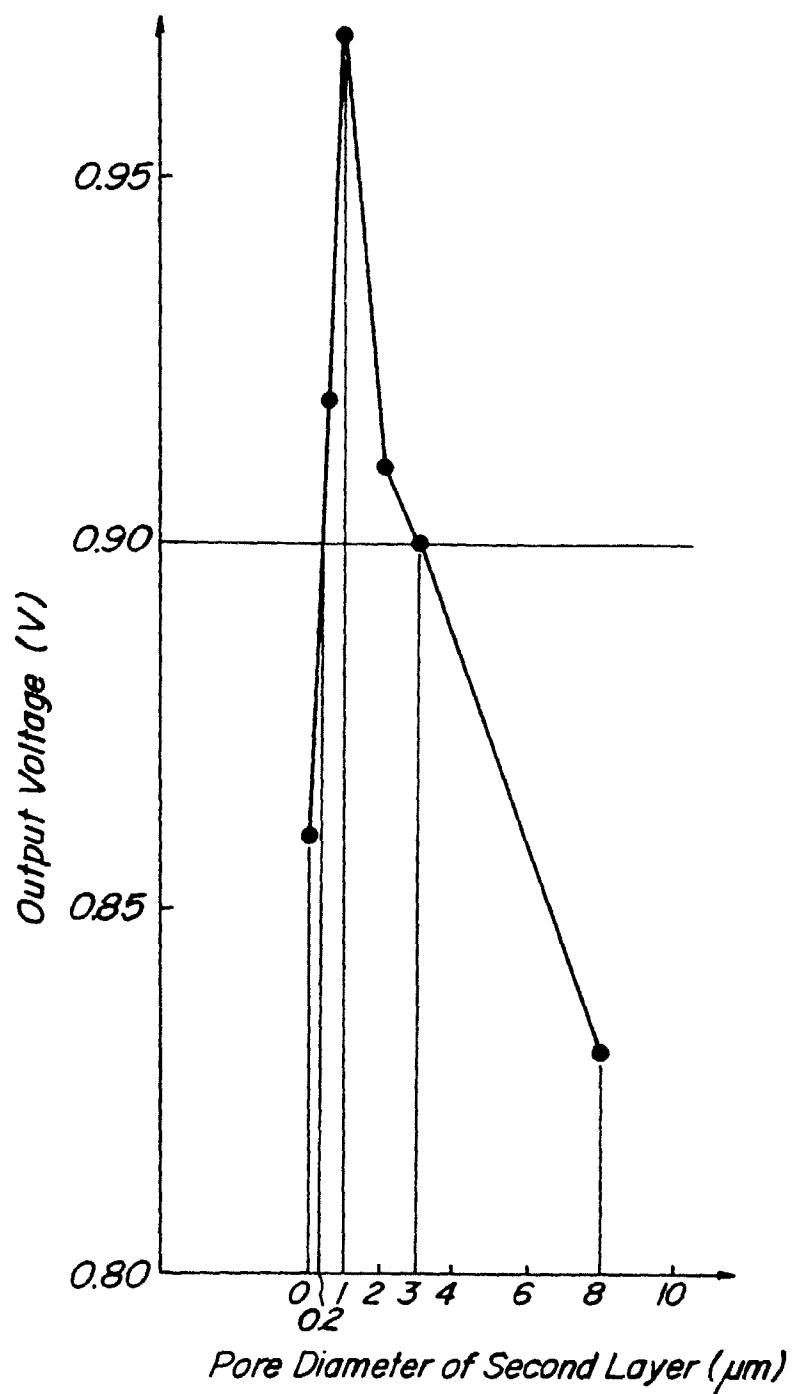


FIG. 8

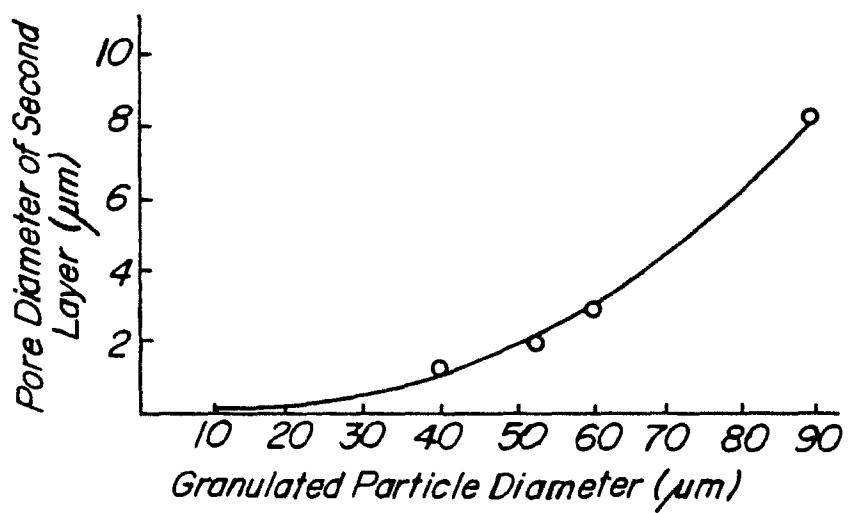


FIG. 9

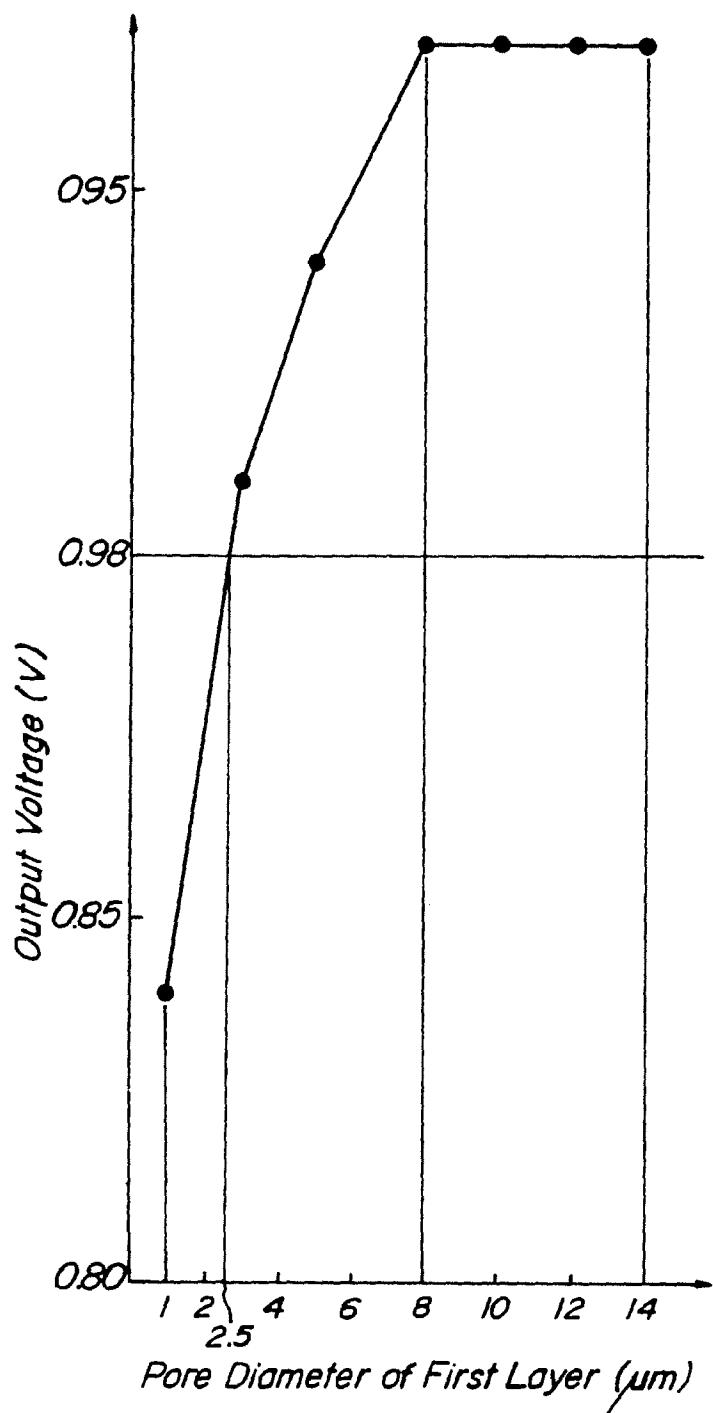


FIG. 10

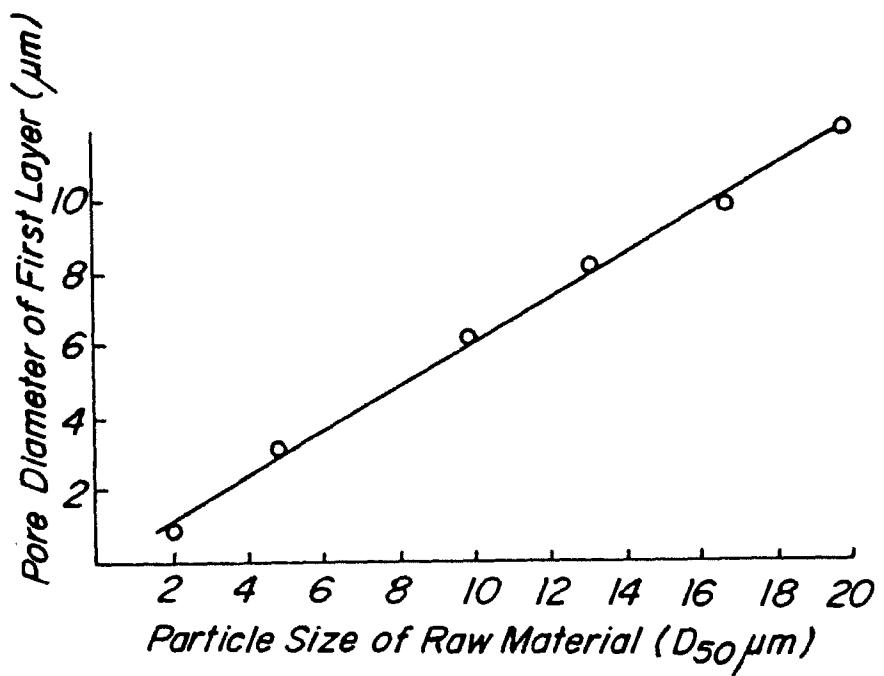


FIG. 11

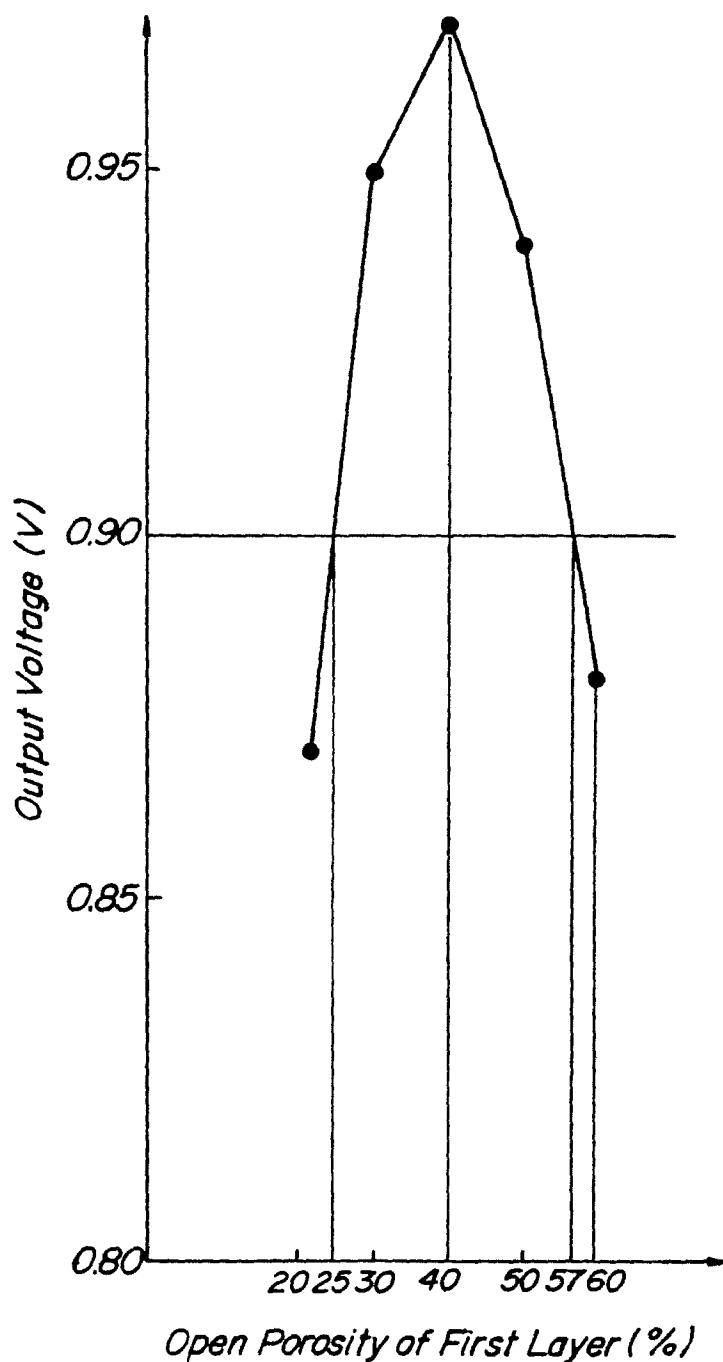


FIG.12

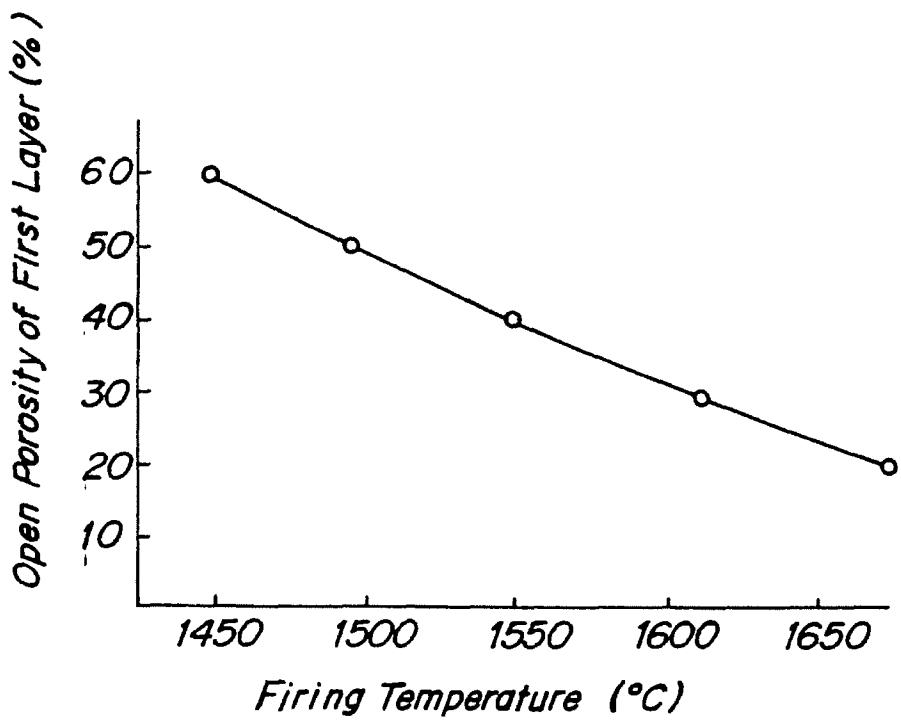


FIG. 13

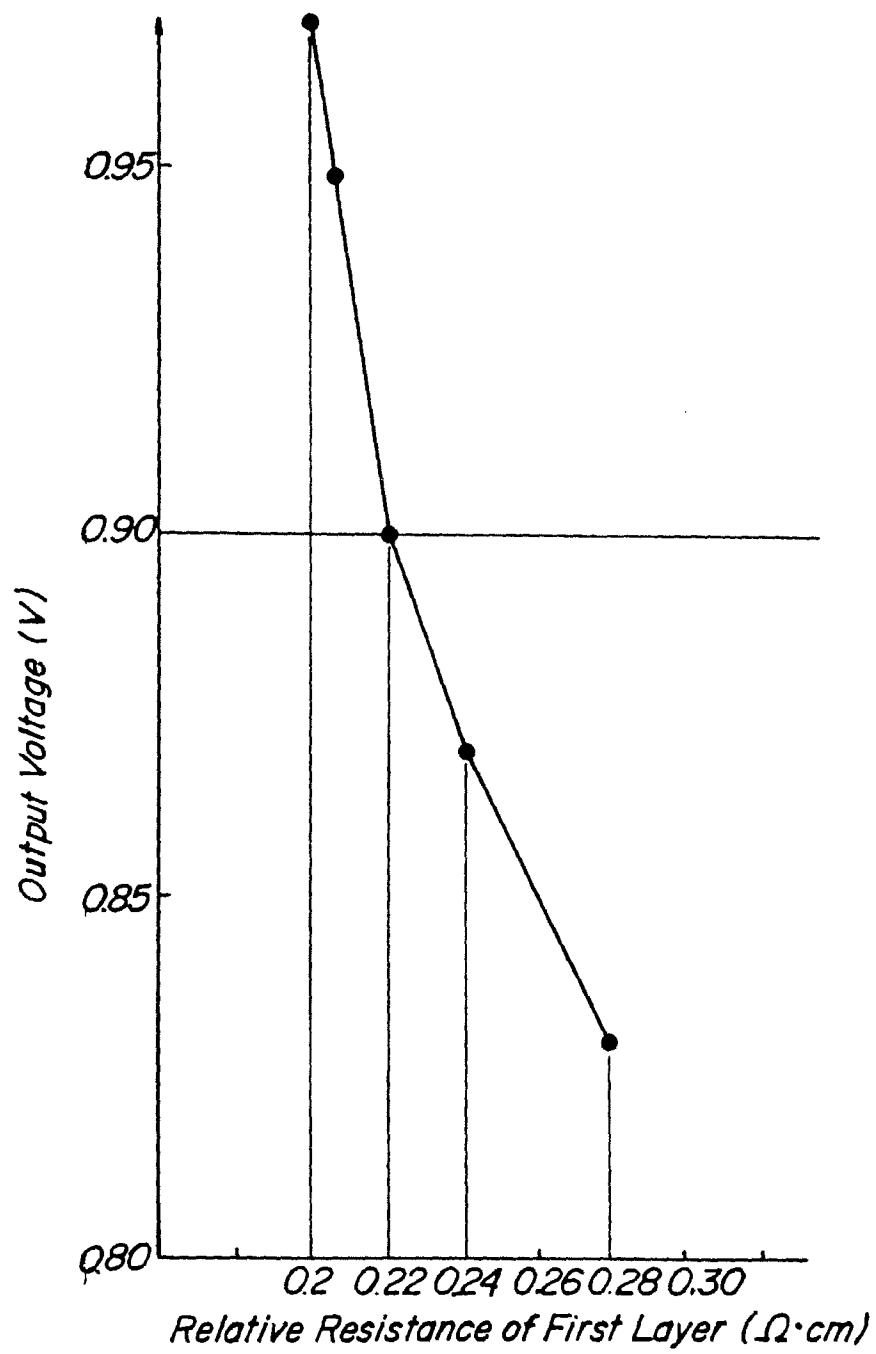


FIG. 14

